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# Two new anthraquinones from *Hedyotis diffusa* W.

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Two new anthraquinones, 2,6-dihydroxy-3-methyl-4-methoxyanthraquinone (1) and 2-hydroxy-7hydroxymethyl-3-methoxyanthraquinone (2), were isolated from *Hedyotis diffusa* W. Their structures were elucidated by means of spectroscopic evidence.

Keywords: Hedyotis diffusa W; Rubiaceae; Anthraquinones; Structure elucidation

# 1. Introduction

*Hedyotis diffusa* W. (Rubiaceae), widely distributed in China, is used as a folk medicine for the treatment of gastroenteritis and appendicitis [1]. Phytochemical investigations on *H. diffusa* have led to the isolation of several classes of compounds such as anthraquinones, iridoids [2] and flavonoids [3]. In our extended research, two new anthraquinones, compounds 1 and 2, were obtained. In this paper, we report the isolation and structural elucidation of the two new anthraquinones.

## 2. Results and discussion

The CHCl<sub>3</sub> parts of the 70% EtOH extract of *H. diffusa* W. were subjected to silica gel column chromatography and RP-HPLC (ODS) to afford two new compounds **1** and **2**. On the basis of spectral data and by comparison with those reported in the literature, the structures of these anthraquinones were established.

Compound **1** was obtained as orange-coloured needles. The molecular formula of  $C_{16}H_{12}O_5$  was determined by EI-MS at m/z 284.0 [M]<sup>+</sup> and HRESI-MS at m/z 307.0584 [M + Na]<sup>+</sup>, which was compatible with NMR analysis. It showed a positive Borntrager's reaction and FeCl<sub>3</sub> reaction, suggesting the presence of an anthraquinone skeleton. The <sup>1</sup>H NMR spectrum of **1** (table 1, figure 1) showed the presence of two  $\beta$ -hydroxyl groups at about  $\delta$  10.90 and

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NO	1					2			
	$\delta_C$	$\delta_H$	НМВС	NOESY	NO	$\delta_C$	$\delta_H$	НМВС	NOESY
1	125.5	7.79 (1H, s)	1a, 2, 9-C		1	112.5	7.54 (1H, s)	1a, 2, 9-C	
1a	124.6				1a	126.7			
2	154.9				2	152.9			
3	131.8				3	152.7			
1	146.9				4	108.7	7.60 (1H, s)	3, 4a, 10-C	$C_3 - OCH_3$
4a	124.2				4a	128.2			
5	112.1	7.42 (1H, d, 2.4 Hz)	5a,10-C		5	126.5	8.12 (1H, d, 8.6 Hz)	5a, 6,10-C	C <sub>6</sub> -H
5a	125.3				5a	131.6			
5	162.8				6	133.1	7.78 (1H, br d, 8.6 Hz)	5, 8-C	C <sub>5</sub> -H, C <sub>7</sub> -CH <sub>2</sub> OH
7	120.9	7.16 (1H, dd, 2.4, 8.5 Hz)	5-C		7	149.4			
3	129.2	7.98 (1H, d, 8.5 Hz)	6, 8a, 9-C		8	123.9	8.10 (1H, br s)	6, 7, 8a, 9-C	C7-CH2OH
8a	136.6				8a	131.9			
9	180.6				9	182.1			
10	181.9				10	181.4			
$C_2 - OH$		10.90*			$C_2$ —OH		5.57		
C <sub>6</sub> —OH		10.10*			C <sub>3</sub> -OCH <sub>3</sub>	56.1	3.98	2-C	C <sub>4</sub> -H
C <sub>3</sub> -CH <sub>3</sub>	16.6	2.28	2, 3-C	C <sub>4</sub> -OCH <sub>3</sub>	C7-CH2OH	62.3	4.68	6, 7, 8-C	C <sub>6</sub> ,C <sub>8</sub> -H
C <sub>4</sub> -OCH <sub>3</sub>	61.2	3.78	4-C	C <sub>3</sub> -CH <sub>3</sub>					

Table 1. NMR (DMSO, 600 MHz) spectral data of 1 and 2.

Two new anthraquinones from H. diffusa

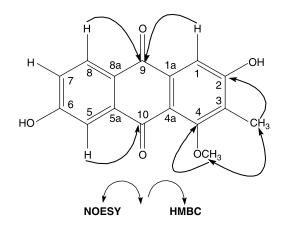


Figure 1. Key HMBC and NOESY correlations of compound 1.

10.10, a methoxyl group at  $\delta$  3.78, a methyl group at  $\delta$  2.28, a proton singlet at  $\delta$  7.79 and three coupled proton signals at  $\delta$  7.98 (1H, d, J = 8.5 Hz, 8-H), 7.16 (1H, dd, J = 8.5, 2.4 Hz, 7-H), 7.42 (1H, d, J = 2.4 Hz, 5-H). The <sup>13</sup>C NMR spectrum (figure 1) revealed 16 carbon signals for two quinone carbonyls, 12 aromatic carbons, a methyl carbon and a methoxyl carbon. A comparison of the <sup>13</sup>C NMR spectral data of **1** with the compound 1,3,6-trihydroxy-2-methylanthraquinone [4] indicated that both compounds have similar structures and the difference was the presence of a methoxyl group in **1** ( $\delta$  3.78, 3H) instead of the hydroxyl group. The HMBC correlations between H-8 ( $\delta$  7.98) and C-9 ( $\delta$  180.6), H-1 ( $\delta$  7.79) and C-9 ( $\delta$  180.6) justified the location of one  $\beta$ -hydroxyl at C-6. The HMBC correlations between H-1 ( $\delta$ 7.79) and C-3 ( $\delta$  131.8), -OCH<sub>3</sub> ( $\delta$  3.78) and C-4 ( $\delta$  146.9), as well as -CH<sub>3</sub> ( $\delta$  2.28) and C-2 ( $\delta$  154.9), C-4 ( $\delta$  146.9) indicated that the methyl group was located at C-3 and the methoxyl group at C-4. In addition, the NOESY spectrum showed a correlation between -OCH<sub>3</sub> ( $\delta$  3.78) and -CH<sub>3</sub> ( $\delta$  2.28) also supported that the methyl group was located at C-3 and the methoxyl group at C-4. And, another  $\beta$ -hydroxyl was located at C-2. Therefore, compound **1** was finally elucidated as 2, 6-dihydroxy-3-methyl-4-methoxyanthra-quinone.

Compound 2 was obtained as orange-coloured needles. Its molecular formula was determined as  $C_{16}H_{12}O_5$  by EI-MS at m/z 284.0 [M]<sup>+</sup> and HRESI-MS at m/z 307.0579  $[M + Na]^+$ . It showed a positive Borntrager's reaction and FeCl<sub>3</sub> reaction, suggesting the presence of an anthraquinone skeleton. The <sup>1</sup>H NMR spectrum of **2** (table 1, figure 2) showed the presence of a  $\beta$ -hydroxyl at  $\delta$  5.57, a methoxyl group at  $\delta$  3.98, a hydroxymethyl group at  $\delta$  4.68, two proton singlets at  $\delta$  7.60 and 7.54, three coupled proton signals at  $\delta$  8.12 (1H, d, J = 8.6 Hz, 5-H), 7.78 (1H, br d, J = 8.6 Hz, 6-H), 8.10 (1H, br s, 8-H). The <sup>13</sup>C NMR spectrum revealed 16 carbon signals for two quinone carbonyls, twelve aromatic carbons, a hydroxymethyl carbon and a methoxyl carbon. Comparison of 2 with a known compound 2-hydroxy-7-methyl-3-methoxy-anthraquinone [5] showed that both compounds have the same substituent pattern, and the difference was the presence of a hydroxymethyl group in compound 2, instead of a methyl group. Furthermore, compound 2 is ascertained as 2-hydroxy-7-hydroxymethyl-3-methoxyanthraquinone or 2-hydroxy-6-hydroxylmethyl-3methoxy-anthraquinone. The proposed structure of 2 and the assignment of NMR spectral data were corroborated by the HMBC and NOESY spectra. The HMBC correlations between H-8 (δ 8.10), H-1 (δ 7.54) with C-9 (δ 181.4), and -CH<sub>2</sub>OH (δ 4.68) with C-6 (δ 133.1), C-7  $(\delta$  149.4), C-8  $(\delta$  123.9) justified the location of the hydroxymethyl group at C-7, and the X.-D. Kang et al.

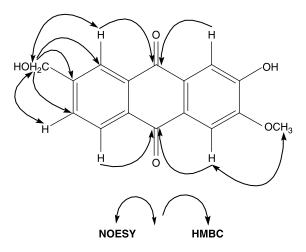


Figure 2. Key HMBC and NOESY correlations of compound 2.

correlation between H-4 ( $\delta$  7.60) and C-10 ( $\delta$  181.4) indicated that the location of the methoxyl was at C-3. In addition, the cross-peak correlation between  $-OCH_3$  ( $\delta$  3.98) and H-4 ( $\delta$  7.60) in the NOESY experiment was observed. These results confirmed the structure of compound **2** as 2-hydroxy-7-hydroxymethyl -3-methoxyanthraquinone.

# 3. Experimental

#### 3.1 General experimental procedures

Melting points were determined on Yanaco MP-S3 melting point apparatus and are uncorrected. The UV spectrum was recorded on a Shimadzu UV-260 UV-Vis instrument. The IR spectrum was recorded on a Bruker IFS-55 instrument. NMR spectra were reported with a Bruker ARX-300 or Bruker ARX-600. EI-MS was performed on VG-5050E mass spectrometer. HRESI-MS was performed on QSTAR LCQ mass spectrometer. RP-HPLC was Shimadzu CTO-6A model with ultraviolet detector.

# 3.2 Plant material

The plant material of *Hedyotis diffusa* W. was collected in Nanchang city, Jiangxi Province, China, in March 2005, and identified by Professor Qishi Sun (Shenyang Pharmaceutical University). A voucher specimen (No. 20050920) is deposited in Research Department of Natural Medicine, Shenyang Pharmaceutical University.

# 3.3 Extraction and isolation

The air-dried herbs (5 kg) of *Hedyotis diffusa* W. were extracted with 70% ethanol (100 L  $\times$  3) for 2 h. Extracts were concentrated *in vacuo* to give a residue (0.8 kg), which was partitioned with petroleum ether, CHCl<sub>3</sub> and n-BuOH successively. The CHCl<sub>3</sub> extract (40 g) was subjected to column chromatographic separation (400 g), gradiently eluted with

petroleum ether/acetone (each 500 ml) to obtain fraction 14 (100:16). Fraction 14 (225 mg) was submitted to preparative thin layer chromatography using petroleum ether/EtOAc/acetone (4:1:1) to afford compound **1** (8 mg) and a mixture containing compound **2** (60 mg). The mixture was separated on RP-HPLC with an ODS column (150 × 4 mm, flow rate 1.0 ml/min) with MeCN/H<sub>2</sub>O (90:10) to yield **2** (10 mg) ( $t_R = 8.5$  min).

**3.3.1 Compound 1**. Orange-coloured needles (MeOH), mp 226–228°C; UV (CH<sub>3</sub>OH)  $\lambda_{\text{max}}$  (log  $\varepsilon$ ) 245 (0.13), 279 (0.50) nm; IR (KBr)  $\nu_{\text{max}}$  3395 (OH), 2925, 2853, 1667 (C=O), 1583, 1474, 1381 cm<sup>-1</sup>. NMR spectral data: see table 1; HRESI-MS: *m/z* 307.0584 (calcd for C<sub>16</sub>H<sub>12</sub>O<sub>5</sub>Na, 307.0582); EI-MS *m/z* 284 [M]<sup>+</sup>(48), 266 [M–H<sub>2</sub>O]<sup>+</sup>(47), 238 [M–H<sub>2</sub>O-CO]<sup>+</sup>(100), 213 [M–CH<sub>3</sub>-2CO]<sup>+</sup>(24), 197 [M–OCH<sub>3</sub>-2CO]<sup>+</sup>(23).

**3.3.2 Compound 2**. Orange-coloured needles (MeOH), mp 223–225°C; UV (CH<sub>3</sub>OH)  $\lambda_{\text{max}}$  (log  $\varepsilon$ ) 246 (0.30), 284 (0.86) nm; IR (KBr)  $\nu_{\text{max}}$  3391 (OH), 2920, 2850, 1668 (C=O), 1590, 1517, 1451 cm<sup>-1</sup>. NMR spectral data: see table 1; HRESI-MS: *m/z* 307.0579 (calcd for C<sub>16</sub>H<sub>12</sub>O<sub>5</sub>Na, 307.0582); EI-MS *m/z* 284 [M]<sup>+</sup>(100), 267 [M–OH]<sup>+</sup>(3), 255 [M–COH]<sup>+</sup>(43), 241 [M–COCH<sub>3</sub>]<sup>+</sup>(11), 225 [M–CO-OCH<sub>3</sub>]<sup>+</sup>(9), 213 [M–CO-COCH<sub>3</sub>]<sup>+</sup>(9).

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